Density functional study on spin-crossover coordination polymers at a molecular level

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1. INTRODUCTION

Spin-crossover (SCO) behaviors of Fe(II) octahedral complexes can be described as the change of ground-state between low spin (t_{2g}^{6}) and high spin $(t_{2g}^4 e_g^2)$. The polymerization of Fe-clusters by bridging ligands has played an important role on controlling whether a SCO phenomenon occurs or not. Coordination polymers (CPs) $[Fe(NCS)_2(L)_2]_n$ (L = 4,4'azopyridine (azpy)¹⁾, 1,2-bis(4-pyridyl)ethane $(bpa)^2$, N^1 , N^2 - bis (pyridine-4-ylmethylene)ethane-1,2-diamine (bpad)³⁾) have shown the switchable properties of SCO-on/off system in a thermal process. Things to be emphasized among them were the systematic difference in Fe-pyridine dihedral angles between SCO-on and SCO-off (Figure). In the present study, we performed a density functional calculation to reveal the spin transition mechanism of SCO-CPs.

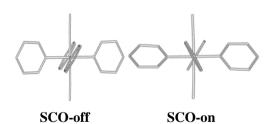


Figure. The structural difference between SCO-on system and SCO-off system.

2. CALCULATION METHOD

All the DFT calculations were performed by using the program "ORCA" package. For the purpose of a molecular-level calculation, a mononuclear pyridine model in which bridging

ligands were replaced by pyridines, [Fe(NCS)₂(pyridine)₄], was created by cutting out a single-crystal X-ray structure. The structure optimization was performed using a BP86/TZVP method by fixing the dihedral angles. All single point calculations were obtained by a TPSSh/TZVP method, which can reproduce the energy difference among different spin-states in the experiments⁴).

3. RESULTS AND DISCUSSION

All-energy difference between high spin and low spin, $\Delta E_0^{\rm HL} = E_0^{\rm HS} - E_0^{\rm LS}$, including all-electron energy and zero-point energy was shown in **Table**. The sign of $\Delta E_0^{\rm HL}$ was reversed depending on whether SCO occurs or not, i.e., the experimental results could be reproduced by our calculation. We will discuss the correlation between a ligand field theory and MOs depending on SCO-on/off.

Table. $\Delta E_0^{\rm HL}$ and SCO-on/off (kcal/mol).

L	guest	$\Delta E_0^{ m ~HL}$	SCO
azpy	No guest	-2.4	off
azpy	EtOH	0.7	on
bpa	No guest	-1.6	off
bpa	<i>p</i> -Dichlorobenzene	1.2	on
bpad	$MeOH + CH_2Cl_2$	-2.6	off
bpad	No guest	0.8	on

4. REFERENCES

- 1) G. J. Halder, et al., Science, 2002, 298, 1762.
- 2) T. Morita, et al., Chem. Lett., 2006, 35, 1042.
- 3) F. Yang, et al., Inorg. Chem., 2011, 50, 911.
- 4) M. Kaneko, et al., Chem. Lett., 2013, 42, 1432.