

# Detection of H atoms by 2-photon LIF technique and Its Application to Chemical Kinetics

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Hydrogen atom is a constituent element of various chemical species and concerned with many reactions. The yields of H atoms enable us to elucidate the mechanisms of chemical reactions. We have constructed a highly sensitive detection system of H atoms and applied it to the kinetic studies on several chemical reactions.

A schematic diagram of the detection system is shown in Fig. 1. It consists of a MgF<sub>2</sub> focusing lens, optical interference filter for the Lyman- $\alpha$  radiation, and VUV photomultiplier tube (PMT). The detection scheme is: (i) H atom in 1s(<sup>1</sup>S) state is excited to 2s(<sup>2</sup>S) state by two photons at 243.2 nm from a dye laser; (ii) The 2s(<sup>2</sup>S) state is transferred to 2p(<sup>2</sup>P) state by collisions with ambient gases; (iii) Lyman- $\alpha$  emission, 2p(<sup>2</sup>P)  $\rightarrow$  1s(<sup>1</sup>S), at 121.6 nm is detected with the PMT. The 2-photon LIF excitation spectrum of H atoms generated in the photolysis of NH<sub>3</sub> at 193 nm is shown in Fig. 2.

Fig. 3 shows the time profiles of H atoms generated in a gaseous mixture O<sub>3</sub>/H<sub>2</sub>/CO/He irradiated with a light at 266 nm. H atoms produced in a reaction OH( $\nu$ ) + CO  $\rightarrow$  H + CO<sub>2</sub> following O(<sup>1</sup>D) + H<sub>2</sub>  $\rightarrow$  OH( $\nu \leq 4$ ) + H were detected, and the time-resolved LIF intensities of H atoms were recorded as a function of the delay times between the photolysis and probe laser. The time profiles of H atoms generated in the reaction OH( $\nu$ ) + CO [Fig. 3(c)] is well-reproduced by simulation with the rate coefficients obtained in the analysis of the time profiles of OH( $\nu = 0 - 4$ ).

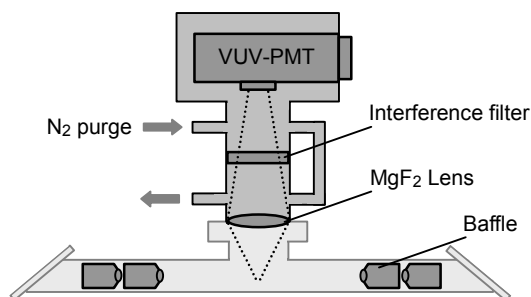


Fig. 1. A schematic diagram of the detection system of H atoms.

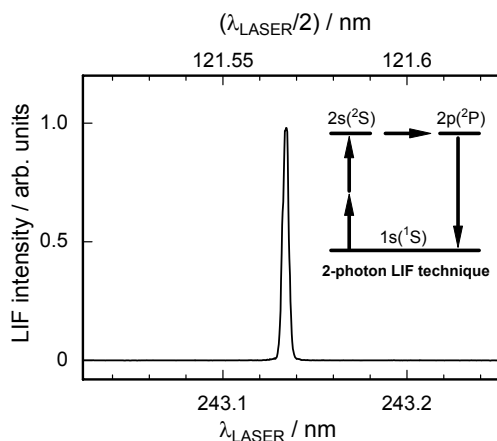


Fig. 2. 2-photon LIF excitation spectrum of H atoms.  $P(\text{NH}_3) = 5$  mTorr,  $P(\text{He}) = 10$  Torr. The inset shows the scheme of detection.

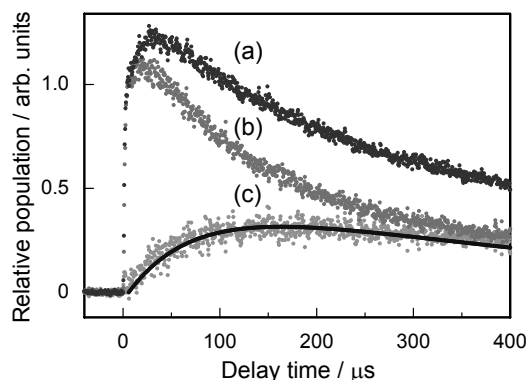


Fig. 3. Time-resolved LIF intensities of H atoms.  $P(\text{O}_3) = 0.2$  mTorr,  $P(\text{H}_2) = 150$  mTorr,  $P(\text{He}) = 10$  Torr. (a)  $P(\text{CO}) = 400$  mTorr and (b)  $P(\text{CO}) = 0$  mTorr. (b) subtracted from (a) leaves (c). The black line denotes the results of simulation.